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Use the periodic table
to identify each
element described
below. 1.

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102–105 Section 4.1

Assessment page 105

1. Contrast the
methods used by the
Greek philos-ophers

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and Dalton to study the
atom. Greek
philosophers could not
conduct

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magnesium is as follows: 79%

magnesium-24, 10%
magnesium-25, and
11% magnesium-26.

Calculate the atomic
mass of magnesium.

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grams of NH₃ (g)
reacted? Is the law of
conservation of mass
observed in the above
reaction? Justify your
answer. 157.5 g 106.5

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g 51 g Yes. Mass of reactants equals mass of products. Section 3.2 Assessment page 79 10. Classify each example as a physical change or a chemical change. a. crushing an aluminum can physical

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a mixture of two
metals to be made in
the world? Answer:

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weight of AgCl will be
precipitated when a
solution containing
4.77 g of NaCl is added
to a solution of 5.77 g
of AgNO_3 ?

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